The Photo-induced Stereospecific Formation of 9-Ethyl-1,4a-dimethyl-4,4a-dihydrocarbazole from *N*-Ethyl-2,6-dimethyldiphenylamine and its Photoreactions

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The light-induced ring closure reaction of N-ethyl-2,6-dimethyldiphenylamine to 9-ethyl-1,4a-dimethyl-4,4a-dihydrocarbazole, (3), and the light-induced ring opening of (3) to give butadienylindoles was found to proceed stereospecifically and in accordance with the Woodward–Hoffmann rules.

N-Ethyl-2,6-dimethyldiphenylamine (1) is converted by light (Scheme 1) into 9-ethyl-1,4a-dimethyl-4,4a-dihydrocarbazole (3) *via* the zwitterionic, transient species (2) which absorbs strongly in the visible region (λ_{max} 640 nm). By means of flash experiments it was shown¹ that a tunnel process contributes significantly to the hydrogen shift reaction (2) \rightarrow (3). In order to calculate quantum mechanical transmission probabilities of the tunnelling proton (H* in Scheme 1) the width of the tunnel barrier had to be determined. Therefore, it was important to find out whether the rearrangement of (2) is regioselective and stereospecific.

Illumination experiments were carried out in methylcyclohexane–isopentane (1:3) under a nitrogen atmosphere at 173 K because the reaction step (2) \rightarrow (3) occurs at this temperature mainly by proton tunnelling. The structure of the primary stable photoproduct (3) was deduced from the 270 MHz FT-n.m.r. spectra† in [2 H₆]DMSO. In order to facilitate the spectrum analysis [and to determine the magnitude of the isotope effect on the kinetics of the reaction (2) \rightarrow (3) 1 J the deuteriated derivative N-ethyl-2',3',4',5',6'-pentadeuterio-2,6-dimethyldiphenylamine (1_D) was also investigated.

The results are as follows. (i), Only the proton H* (cf. Scheme I) shifts from the phenyl moiety of (1) via the transient (2) to the methyl-substituted ring of (3). In the case of (1_D) , which yields (3_D) , the proton H* at C-4 of (3) $(\delta 2.25)$ is replaced by a deuteron and the signal at $\delta 2.25$ is entirely missing.

Scheme 1

The signal of the second proton at C-4 (δ 2.48) appears in the spectrum of (3) and of (3_D) with the same magnitude. The other four ²H-atoms remain in the phenyl ring of (3_D). (ii), Only a [1,8]-sigmatropic proton shift takes place. (A [1,4]-shift is also conceivable, but thermodynamically less favourable. (iii), The methyl group at the bridge of (3) (carbon atom C-4a) and the shifted proton H*at C-4 are in a *trans*-configuration. This follows from the coupling constants, J, of the

^{† &}lt;sup>1</sup>H n.m.r. data for (3) measured in [2H_6]dimethyl sulphoxide at 300 K: δ 1.00 (3H, s, 4a-CH₃); 1.14 (3H,t, J 6.8 Hz, ethyl CH₃), 1.94 (3H,s,1-CH₃), 2.25 (1H, dd, $J_{3.4}^*$ 3, $J_{4.4}^*$ 16.5, and $J_{2.4}^*$ 3 Hz, 4-H*), 2.48 (1H, dd, $J_{3.4}$ 6.5Hz, 4-H), 3.81 (2H, m, ethyl CH₂), 5.30 (1H, m, 3-H), 5.78 (1H, dd, $J_{2.3}$ 10 Hz, 2-H), and 6.6—7.2 (4H,m, aromatic H).

(3)
$$\frac{hv}{conrot}$$
.

 R^1
 R^2
 R^1
 R^2
 R^1
 R^2
 R^3
 R^4
 R^4

geminal protons H and H* at C-4 of (3) with the proton at C-3; $J_{3,4}$ 6.5 and $J_{3,4}$ * < 3 Hz. Molecular models of (3) show that the conformation of these three protons is rigidly fixed. If the CH₃-group at C-4a and H* are in a *trans*-relationship, the dihedral angles between the proton at C-3 and the protons H (δ 2.48) and H* (δ 2.25) at C-4 are about 30 and 90°, respectively. The observed coupling constants are in agreement with the values predicted from the Karplus-Conroy curve for such angles. [In ($\mathbf{3}_{\rm D}$) only the coupling $J_{3,4}$ 6.5 Hz is observed.]

The structure of the transient (2) could not be determined experimentally because of its short life time. However, the Woodward-Hoffmann (WH) rules predict that in an electronically excited state the 1,5-electrocyclic ring closure reaction (1) \rightarrow (2) takes place conrotatorily. If this is the case the CH₃-group and the proton H* at the bridge of (2) are in the trans-configuration and the (thermal) [1,8]-sigmatropic proton shift (2) \rightarrow (3) is, in agreement with the WH rules, suprafacial.

The photoreactions of (3) and (3_D) demonstrate the applicability of the WH rules to the entire reaction sequence. As in other cyclohexadiene systems (e.g. 1,2-dihydronaphthalenes³ or ergosterol⁴) the ring opening reaction (3) \rightarrow (4) (Scheme 2) is stereospecific. The WH rules predict a conrotatory ring opening for the light-induced reaction. If one avoids consecutive photoreactions as much as possible by photolysing only about 50% of (1), one observes the formation of (3) and of the stereoisomers 1-ethyl-3-methyl-2-(cis-1'-methylbuta-1',3'-dienyl)indole (4) and 1-ethyl-3-methyl-2-(cis-1'-methylbuta-1',3'-dienyl)indole (5) with the concentration ratio for (3):(4):(5) of 15:6:1.

Using (1_D) one can distinguish between the n.m.r. spectra of the isomers (4a), (4b) and (5a), (5b) (cf. Scheme 2). Under the above conditions (50% conversion) the concentration ratio (4a): (4b) was at least 10:1 (n.m.r. detection limit). The concentration of (5) was too small to determine the concentration ratio (5a): (5b) accurately. Higher conversions of (1) or (1_D) yield mainly (4) and (5) in a concentration ratio (4): (5) of (5): (5) and very little (3).

The concentration ratios (4a): (4b) and (5a): (5b) are in this case 7:3. It is possible to separate (4) and (5) by means of gas chromatography. After separation of the deuteriated compounds the concentration ratios (4a): (4b) and (5a): (5b) are still 7:3. Illumination of the mixture (4a), (4b) leads to a photostationary state mixture of isomers with the concentration ratios for (4): (5) of 3:2 and for (4a): (4b) and (5a): (5b) of 1:1.

We conclude from these results that the first step of the ring opening reaction is indeed stereospecific, yielding exclusively (4a) as a result of a conrotatory reaction in accordance with the WH rules. The three other *cis-trans*-isomers are the products of consecutive photoreactions.

Received, 22nd February 1982; Com. 186

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